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forces: attractive

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between all molecules  
and atoms B.  
Intermolecular forces  
are responsible for the  
very existence of  
condensed states C.  
States of liquid, solid,  
or gas are based on  
magnitude of  
intermolecular forces  
II. 11.2 Solids, Liquids,  
and Gases: A Molecular  
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Chapter 11/4b FAQ. Q:  
What are hydrophobic  
and hydrophilic? A:  
Hydrophobic is a “fear  
of water”. Nonpolar  
substances are  
typically hydrophobic.  
They are not attracted  
to water. Hydrophilic  
on the other hand are  
“water lovers”. These  
compounds are  
attracted to water, and  
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close to the same, they  
SHARE electrons.

Bonds formed when  
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Bonds .

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forces that exist  
between all molecules  
and atoms B.  
Intermolecular forces  
are responsible for the  
very existence of  
condensed states C.  
States of liquid, solid,  
or gas are based on  
magnitude of  
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hydrophilic? A:  
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of water”. Nonpolar  
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gases. equal volumes  
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pressure contain equal numbers of molecules. What is Avogadro's standard molar volume for gases at STP?

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